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Preliminary communication

Oligomerization of butadiene with $(\eta - C_5 R_5) Ru^{II}$ complexes; stoichiometric and catalytic chemistry *

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Abstract

Treatment of $(\eta$ -C₅R₅)Ru(η -butadiene)X; (R = H, Me; X = Br, Cl) with butadiene, in the presence of silver triflate, yielded cationic complexes with 1,3,7-octatriene ligands, produced from the C-C bond formation at both the terminal carbon atoms of two molecules of butadiene. In the pentamethylcyclopentadienyl system the octatriene ligand was cyclodimerized to 1,5-cyclooctadiene when treated with carbon monoxide. The ligand structure was elucidated by NMR spectroscopy. Catalytic reactions on these systems have also been studied.

There have been a number of studies on catalytic oligomerization of conjugated dienes by use of a wide range of transition-metal complexes [1]. In these reactions, regio- and stereo-selective formation of the desired products was crucially important. We report here stoichiometric and catalytic oligomerization of butadiene with high selectivity by use of " $(\eta - C_5 R_5) Ru^+$ " species.

The complex $(\eta$ -C₅R₅)Ru(η -butadiene)Br (1, 2a) was easily prepared by reducing $(\eta$ -C₅R₅)RuBr₂(η -C₃H₅) [2], with a 4-6 fold excess of triethylaluminium, in the presence of butadiene under mild conditions (eq. 1, Yield: Cp, R = H, 83%; Cp^{*}, R = Me, 97.8%). Alternatively Cp^{*}Ru(η -butadiene)Cl (2b) was also prepared directly from the treatment of [Cp^{*}RuCl₂]_n [3] with butadiene in the presence of zinc powder in methanol at 0°C (71% yield, eq. 2). These Cp^{*}Ru(η -butadiene)X complexes 2 showed similar *exo*-conformations to those reported by Fagan *et al.* [4], as evidenced by their ¹H and ¹³C NMR spectra.

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When CpRu(η -butadiene)Br (1, 30 mg, 0.1 mmol) was treated with an excess of butadiene (0.5 mmol) and silver triflate (0.12 mmol) in dichloromethane (8 mL) at -30° C for 1 h, pale yellow [CpRu(C₄H₆)₂]OTf (3, OTf = trifluoromethanesulfonate) was obtained in 78% yield. Similarly, the treatment of Cp*Ru(η -butadiene)Br (2a) with butadiene and silver triflate gave the 1,3,7-octatriene complex 4 (95% yield) that was thermally more labile around room temperature than its cyclopentadienyl analogue 3.

The structure of the octatriene ligands in these complexes was carefully examined by ¹H and ¹³C NMR spectroscopy [5*]. On the bases of the relative intensities of signals to the Cp or Cp* ligands, and supported by ¹³C NMR spectra, it could be concluded that the coordinated ligands in both complexes have the C_8H_{12} composition. Existence of eight olefinic and four methylenic proton signals strongly indicated the ligand to be 1,3,7-octatriene, formed by C–C bond formation at both the terminal carbon atoms of two butadiene molecules, followed by regioselective hydrogen atom migration.

This Cp^{*} octatriene complex (4) converted into the 1,5-cyclooctadiene complex (5) upon exposure to carbon monoxide (1 atm). [Cp^{*}Ru⁺(η -COD)CO]OTf (5) (COD = 1,5-cyclooctadiene) was identified by comparing spectral data with those of the authentic product independently prepared by the reaction of Cp^{*}Ru(η -COD)Cl and CO in the presence of AgOTf [6^{*}]. Thus, 1,5-cyclooctadiene, coordinated to ruthenium was formally generated by [$\pi 4 + \pi 4$] cycloaddition between two butadiene molecules (Scheme 1).

However, the octatriene ligand of the Cp complex 3 could not be cyclized in the presence of carbon monoxide (1 atm) into the 1,5-COD ligand. This indicates that the octatriene ligand in 4 has a different configuration from that in 3.

In fact, the ¹H NMR of 4 appeared as a triplet signal at δ 3.30 (J = 8.3), assignable to the proton at C₃. Identical proton coupling constants of J_{2-3} and J_{3-4} strongly suggest that the olefinic geometry at the C₃-C₄ bond has (Z) configuration.

In contrast, the ¹H NMR spectra of 3 showed a double doublet signal of proton attached to the C₃ carbon at δ 2.95 ($J_{2-3} = 7.9$; $J_{3-4} = 11.6$ Hz). This suggests the 1,3,7-octatriene ligand in 3 being the more thermodynamically stable (*E*) isomer around the C₃-C₄ bond. In the case of 3, the configuration puts the terminal

^{*} Reference number with asterisk indicates a note in the list of references.



Scheme 1

carbon atoms (C_1 and C_8) too far from each other to induce the second step C-C bond formation to the 1,5-COD ligand.

Catalytic oligomerization of butadiene was studied by using these cationic complexes as catalyst. Results are included in Scheme 1. In the Cp^{*} system, the best result was obtained when 1 mol% of $[Cp^*Ru(\eta-C_4H_6)]OTf$ and butadiene in THF were used at 70°C for 10 h; 1,5-cyclooctadiene was selectively obtained (89%; Ru based yield was 4440%).

On the other hand, the catalytic reaction with 1 mol% of [CpRu(1-4; 7-8- η -C₈H₁₂)]TfO (3), at 60°C for 4 days in dichloromethane, resulted in the formation of linear trimer products (C12 over 92% selectivity), contaminated with a little C8 and C16 linear oligomers. The linear trimers were a 55:45 mixture of 1,5,7,11-dodecatetraene and its 1,3,7,11-isomer.

The first step involved was the treatment of butadiene complexes 3 and 4 with silver triflate, generating a cationic coordinatively unsaturated intermediate, at which the coordination of the second butadiene took place. Two coordinated butadiene molecules induced the C-C bond formation between the terminal carbon atoms to give bis(allyl)Ru^{1V} intermediates. Rapid hydride migration in the bis(allyl) complex gave 1,3,7-octatriene Ru^{II} complexes. In the case of the Cp system, further insertion of butadiene between the diene part of the (*E*)-1,3,7-octatriene ligand in 3 and the third molecule of butadiene took place.

Alternatively, as the distance between C1 and C8 in the (Z)-1,3,7-octatriene ligand in the Cp^{*} system (4) is extremely close, the coordination of further butadiene induced spontaneous cyclization of the (Z)-octatriene ligand, and the 1,5-COD is freed to regenerate the active $[Cp^*Ru(\eta-C_4H_6)]^+$ species.

References and notes

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- 5 Spectral data. [CpRu(η-C₈H₁₂)]TfO: ¹H-NMR (270 MHz, acetone-d₆): 1.39 (1H, m), 2.04 (1H, m), 2.53 (1H, d), 2.64–2.78 (1H, m), 2.86–2.99 (1H, m), 2.95 (1H, dd), 3.92 (1H, d), 4.01 (1H, d), 4.05 (1H, d), 4.71 (1H, m), 5.70 (5H, s). ¹³C-NMR (67.8 MHz, acetone-d₆): 29.7, 40.8, 50.3, 56.6, 81.6, 83.6, 85.5, 91.5 (ring), 98.1. Elemental analysis. Found: C, 38.96; H, 3.91. C₁₄H₁₇F₃SO₃Ru calcd.: C, 39.71; H, 4.05%. Melting point 152–153°C (dec.). [Cp*Ru(η-C₈H₁₂)]TfO: ¹H-NMR (270 MHz, CDCl₃): 1.77 (15H, s), 2.01 (2H, d), 2.52 (1H, d), 2.71

(1H, m), 3.01 (2H, m), 3.30 (1H, t), 3.33 (1H, m), 3.85 (1H, m), 4.26 (1H, m), 4.27 (1H, d), 5.01 (1H, m). ^{13}C -NMR (67.8 MHz, CDCl₃, $-10^{\circ}C$): 9.65, 37.2, 38.7, 62.5, 67.7, 90.2, 90.3, 98.1, 102.1, 105.0.

6 Pale yellow crystals in 62% yield. Spectral data: [Cp*Ru(η-1,5-COD)CO]TfO: ¹H-NMR (270 MHz, acetone-d₆): 1.94 (15H, s), 2.29 (4H, m), 2.58 (4H, m), 4.25 (2H, m), 4.47 (2H, m). ¹³C-NMR (67.8 MHz, acetone-d₆): 10.3, 29.6, 30.5, 89.0, 93.4, 100.9, 206.8.